

Isoflavonoid Phytoalexins from Leaves of *Trifolium arvense*

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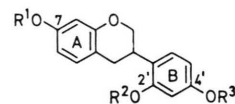
Leguminosae, *Trifolium*, Isoflavans, Pterocarpan,
Phytoalexins

In response to fungal inoculation the leaves of *Trifolium arvense* accumulate the known isoflavonoid phytoalexins, medicarpin, maackiain, sativan, vestitol, and isovestitol. A previously undescribed isoflavan derivative (arvensan; 7,2'-dimethoxy-4'-hydroxyisoflavan) is also produced by this species. The identification and synthesis of arvensan is described.

Numerous isoflavonoid phytoalexins¹ have been isolated from species belonging to the family Leguminosae (subfamily Lotoideae) and implicated as factors in disease resistance^{2,3}. These compounds are often pterocarpin or isoflavan in nature^{2,4,5} although isoflavone⁶ and isoflavanone^{6,7} phytoalexins have also been described. A recent comprehensive survey of genera (e.g. *Melilotus*, *Medicago*, *Trigonella*, and *Trifolium*) within the tribe Trifolieae has revealed that the leaves of many species accumulate isoflavan and pterocarpin derivatives following inoculation with conidial suspensions of the fungus, *Helminthosporium carbonum* Ullstrup^{3,4}. This paper reports the isolation of a new phytoalexin from hare's-foot clover (*Trifolium arvense* L.) and presents evidence to support its identification as 7,2'-dimethoxy-4'-hydroxyisoflavan (**1**).

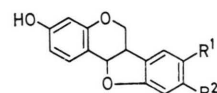
Phytoalexins were obtained from the *H. carbonum*-inoculated leaflets of *T. arvense* using the drop-diffusate technique⁵. TLC (CHCl₃ : MeOH, 50 : 1) of diffusate extracts^{4,5} gave two broad phenolic bands at *R_F* 0.59–0.66 (termed B-1) and *R_F* 0.43–0.47 (B-2). These were eluted with EtOH and the solvent removed *in vacuo* (40 °C). The B-1 fraction was then further chromatographed (CHCl₃, ×5) to afford, i) a mixture of medicarpin and maackiain (**2**) and (**3**) (lower zone), ii) sativan (**4**) (intermediate zone) and iii) an upper band (compound **1**) which gave a bright orange coloura-

tion when sprayed with diazotised *p*-nitroaniline; compounds **2–4** are all characterised by the intense yellow colour of their diazo derivatives. Medicarpin and maackiain were eventually resolved by TLC (×3) in *n*-pentane : Et₂O : HOAc (75 : 25 : 3)⁴.



- 1:** R¹=R²=CH₃; R³=H
4: R¹=H; R²=R³=CH₃
5: R¹=R²=H; R³=CH₃
6: R¹=R³=H; R²=CH₃
7: R¹=R³=CH₃; R²=H
8: R¹=R²=R³=CH₃

Upon multiple development (×3) in this latter system, B-2 also separated into two components namely, vestitol (**5**) (upper zone) and isovestitol (**6**) (lower zone). All the above compounds were chromatographically homogeneous and were absent from extracts of the control leaf diffusate⁵. The identification of compounds **2–6** was based on a UV and TLC (5 solvent systems) comparison with authentic material and on the colours formed after spraying developed chromatograms with Gibbs reagent⁸ and diazotised *p*-nitroaniline.



- 2:** R¹=H; R²=OCH₃
3: R¹=R²=O-CH₂-O

Although medicarpin, vestitol and sativan are common as phytoalexins in members of the tribe Trifolieae³, maackiain is encountered less frequently and as yet has been associated with only two genera (*Trigonella* and *Trifolium*^{4,9,10}) of this tribe. Isovestitol has previously been described as a phytoalexin of *Anthyllis vulneraria* and five *Tetragonolobus* species¹¹ (tribe Loteae) but in the Trifolieae it is of exceptionally rare occurrence³. However, in addition to *T. arvense*, isovestitol has recently been isolated (together with **2**, **3**, **4**, and **5**) from *T. rubens*³. There was no evidence to suggest that *T. arvense* produced either 4-methoxymaackiain or isosativan (**7**), two isoflavonoid phytoalexins obtained from the leaves of alsike clover (*T. hybridum*)⁴.

Compound **1** exhibited a mass spectral fragmentation pattern entirely consistent with its identification as a trisubstituted (dimethoxy-monohydroxy)

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isoflavan^{12, 13}. In fact, the mass and UV (EtOH) spectral data recorded for **1** (see Experimental) closely resembled those of 7,4'-dimethoxy-2'-hydroxyisoflavan **7** (isosativan)⁴. The molecular ion was evident at m/e 286 and was accompanied by three major fragments at m/e 150 (*a*), 149 (*b*) and 137 (*c*; base peak). These ions allow the aromatic ring substituents (which biogenetically should be at C-7, 2' and 4') to be assigned as follows, i) *A-RING*: C-7 (OCH₃; fragment *b*) and ii) *B-RING*: C-2'/C-4' (OCH₃/OH; fragment *a* and *c*). As **1** did not react to Gibbs reagent^{8, 14} (*cf.* isosativan, deep blue⁴) the single B-ring hydroxyl group must be located at C-4' rather than at C-2'. The above oxygenation pattern was confirmed by methylation (with diazomethane¹⁵) to afford a monomethyl ether identical (UV, MS, TLC) with 7,2',4'-trimethoxyisoflavan (**8**) similarly prepared from vestitol (**5**). Compound **1** is thus 7,2'-dimethoxy-4'-hydroxyisoflavan; this substance (which has not previously been described as either a natural or synthetic product) has been named *arvensan* after the source plant, *Trifolium arvense*.

Arvensan has been synthesised *via* 7,4'-dibenzyl-oxy-2'-methoxyisoflavone¹⁶ itself produced from the corresponding 2'-acetoxychalcone by oxidative rearrangement using thallium (III) nitrate trihydrate¹⁷. Debenzylation of the above isoflavone followed by selective methylation and hydrogenation gave a product chromatographically and spectrally indistinguishable from natural arvensan. As noted earlier, arvensan gives a deep orange colouration when sprayed with diazotised *p*-nitroaniline. This feature is characteristic of several isoflavans^{3, 18} (*e.g.* **6**) which possess at C-4' hydroxyl group. In contrast, compounds with methoxylation at C-4' often give a bright yellow colour (*e.g.* **4** and **5**) with diazotised *p*-nitroaniline.

Im TLC bioassays against spore germination of *Cladosporium herbarum* Fr.⁸ pronounced antifungal zones were associated with applied arvensan levels of 10, 20, and 30 μg . When tested against the mycelial growth of *H. carbonum*⁸ arvensan had an ED₅₀ value of about 30 $\mu\text{g}/\text{ml}$; its fungitoxic activity is thus of approximately the same order as that of medicarpin (ED₅₀ 25 $\mu\text{g}/\text{ml}$) and maackiain (ED₅₀ 33 $\mu\text{g}/\text{ml}$)⁸. The other isoflavans from *T. arvense* appear slightly more inhibitory than arvensan (*cf.* **4**, ED₅₀ 10 $\mu\text{g}/\text{ml}$; **5**, ED₅₀ 17 $\mu\text{g}/\text{ml}$; **6**, ED₅₀ 23 $\mu\text{g}/\text{ml}$). Nevertheless, the high arvensan level (50 $\mu\text{g}/\text{ml}$) associated with *H. carbonum*-induced diffusates suggests that formation of this compound may contribute significantly to disease resistance. Substantial quantities of the other isoflavonoid phytoalexins (**2–6**) were also produced

by *T. arvense* (**2**, 65 $\mu\text{g}/\text{ml}$; **3**, 63 $\mu\text{g}/\text{ml}$; **4**, 105 $\mu\text{g}/\text{ml}$; **5**, 65 $\mu\text{g}/\text{ml}$; **6**, 123 $\mu\text{g}/\text{ml}$).

The formation, by *T. arvense*, of several structurally similar pterocarpan and isoflavans suggests that in the plant these compounds may be biosynthetically related. In *Medicago sativa* there is labelling evidence to support the view that sativan is derived *via* methylation of vestitol and that vestitol and medicarpin can be interconverted¹⁹. However, the latter two compounds may originate simultaneously from a common intermediate¹⁹. Although arvensan may be derived by methylation of isovestitol, no logical demethyl precursor of the latter compound was isolated from *T. arvense*. In the genus *Tetragonolobus*, isovestitol co-occurs with 7,2',4'-trihydroxyisoflavan (demethylvestitol¹¹), a substance which could be readily converted to compound **5** by methylation at C-2'. However, repeated investigation has failed to reveal the production of demethylvestitol by *T. arvense*; it is conceivable, therefore, that isovestitol may originate by host-plant demethylation of sativan. No evidence was obtained to suggest that *T. arvense* accumulated methylenedioxy substituted isoflavans related to maackiain.

Experimental

Mass and UV spectra were determined as previously described⁶. TLC separations were undertaken using precoated, glass-backed plates (Merck Kieselgel 60 F₂₅₄, layer thickness, 0.25 mm).

Plant material. Seeds of *Trifolium arvense* L. (obtained from the Botanic Garden, Klagenfurt, Austria) were grown as previously described²⁰. Leaflets for fungal inoculation⁵ were collected when the plants were 6–9 months old.

Fungal material. Cultural conditions, preparation of spore suspensions and source of *Helminthosporium carbonum* Ullstrup have been reported elsewhere⁸.

7,2'-Dimethoxy-4'-hydroxyisoflavan (**1**) (*arvensan*). Colour with diazotised *p*-nitroaniline, orange; no colour was observed with Gibbs reagent. λ_{max} EtOH (nm) 205 (log ϵ 4.70), 225 (4.17), 280–282 (3.77), 287sh (3.71); EtOH + NaOH (nm) 217, 244sh, 285, 290, 298sh; MS rel. int.) 287 (15), 286(M⁺; 89), 151(40), 150(84), 149(35), 148(30), 138(27), 137(100), 135(42), 121(45), 107(41). Monomethyl ether **8** (R_F 0.93, CHCl₃) UV and MS as lit.^{4, 11}. Monoacetate (R_F 0.16, CCl₄:CHCl₃, 4:1); λ_{max} EtOH (nm) 213, 225sh, 275sh, 280, 289sh; MS (rel. int.) 329(6), 328(M⁺; 40), 286(23), 192(21), 151(11), 150(100), 149(42), 148(22), 138(13), 137(52), 135(29), 121(14), 119(11), 107(18).

Synthesis of arvensan. a) 7,4'-Dihydroxy-2'-methoxyisoflavone. This intermediate was synthesised as previously described¹⁷.

b) 7,2'-Dimethoxy-4'-hydroxyisoflavone. The preceding isoflavone (150 mg in DMF (20 ml) was stirred (60 °C) for 1 h with K₂CO₃ (2 g) and CH₃I (75 mg). The mixture was then poured into H₂O and extracted with EtOAc (×6). After removal of EtOAc (*in vacuo*, 40 °C) the residue was crystallised from MeOH to give the desired product (108 mg). λ_{\max} MeOH (nm) 211, 240, 248, 287, 306sh; MeOH + NaOH (nm) 211, 240, 247sh, 281, 291sh, 305sh; MS (rel. int.) 299(23), 298 (M⁺; 100), 297(18), 281(15), 269(10), 268(11), 267(65), 152(6), 151(77), 149(13), 148(10), 147(27), 146(7), 119(9), 107(6), 105(7); mp. 212–217 °C. Colour with diazotised *p*-nitroaniline, orange.

c) (±)-7,7'-Dimethoxy-4'-hydroxyisoflavon (1). The above isoflavone (87 mg), HOAc (20 ml) and Pd-C (10%; 80 mg) were shaken overnight in an atmosphere of H₂. After filtration and removal of solvent, the product was chromatographed (C₆H₆: EtOAc: MeOH: petrol, 6:4:1:3) to give an oil which solidified upon treatment with a little aq. MeOH (yield, 55 mg). UV and MS as reported for the natural product; mp. 111–115 °C. The synthetic and natural isoflavans had identical *R_F* values in CHCl₃: MeOH (50:1, *R_F* 0.65), C₆H₆: MeOH (9:1, *R_F* 0.80), CHCl₃ (*R_F* 0.20), Et₂O: *n*-hexane (3:1, *R_F* 0.83), and *n*-pentane: Et₂O: HOAc (75:25:1, *R_F* 0.48).

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- ¹ J. L. Ingham, *Botan. Rev.* **38**, 343 [1972].
- ² H. D. VanEtten and S. G. Pueppke, *Biochemical Aspects of Plant-Parasite Relationships* (J. Friend and D. R. Threlfall, ed.), p. 239, Academic Press, London 1976.
- ³ J. L. Ingham, Ph. D. thesis, University of Reading, U.K. 1976.
- ⁴ J. L. Ingham, *Z. Naturforsch.* **31c**, 331 [1976].
- ⁵ J. L. Ingham and R. L. Millar, *Nature* **242**, 125 [1973].
- ⁶ J. L. Ingham, *Z. Naturforsch.* **31c**, 504 [1976].
- ⁷ R. S. Burden, J. A. Bailey, and G. W. Dawson, *Tetrahedron Lett.* **1972**, 4175.
- ⁸ J. L. Ingham, *Phytopathol. Z.* **87**, 353 [1976].
- ⁹ V. J. Higgins and D. G. Smith, *Phytopathology* **62**, 235 [1972].
- ¹⁰ J. L. Ingham and J. B. Harborne, *Nature* **260**, 241 [1976].
- ¹¹ J. L. Ingham, *Phytochemistry*, in press.
- ¹² A. Pelter, P. Stainton, and M. Barber, *J. Heterocyclic Chem.* **2**, 262 [1965].
- ¹³ A. Pelter and P. I. Amenechi, *J. Chem. Soc., C* **1969**, 887.
- ¹⁴ F. E. King, T. J. King, and L. C. Manning, *J. Chem. Soc.* **1957**, 563.
- ¹⁵ L. E. Powell, *Plant Physiol.* **39**, 836 [1964].
- ¹⁶ P. M. Dewick, *Phytochemistry*, in press.
- ¹⁷ L. Farkas, Á. Gottsegen, M. Nógrádi, and S. Antus, *J. Chem. Soc. Perkin I* **1974**, 305.
- ¹⁸ N. W. Preston, *Phytochemistry* **14**, 1131 [1975].
- ¹⁹ P. M. Dewick and M. Martin, *J. Chem. Soc. Chem. Commun.* **1976**, 637.
- ²⁰ J. L. Ingham, *Phytochemistry* **15**, 1489 [1976].